

## Maden Tetkik ve Arama Dergisi

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### Mineral chemistry, petrography and crystallization conditions of the Middle Eocene Kazıkbeli Pluton (Eastern Pontides, NE Türkiye)

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Research Article

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#### ABSTRACT

The Eastern Pontides host a diverse suite of plutonic rocks spanning a wide range of ages and compositions. Among these, the Middle Eocene Kazıkbeli pluton, located in the Kürtün district of Gümüşhane, stands out due to its distinctive petrological characteristics. This study aims to unravel the petrological implications of petrographic and mineral chemical data to constrain the physicayduncakirochemical conditions (temperature, pressure, oxygen fugacity) under which the Kazikbeli magma crystallized and was emplaced. By integrating mineral chemical data, we seek to quantify emplacement pressure, crystallization temperature, and oxygen fugacity. A comprehensive understanding of the genetic relationships and physicochemical properties of the Kazıkbeli pluton rocks, as determined through geological, petrographic and mineral chemistry, is crucial for elucidating the geological evolution of the Eastern Pontides. The Kazıkbeli Pluton exhibits a predominant NE-SW orientation and encompasses an area of roughly 46 km<sup>2</sup>. Modal mineralogical analysis reveals a compositional spectrum ranging from gabbroic diorite to monzogranite, with granodiorite and tonalite being the most prominent rock types. Textural variations encompass fine- to medium-grained, porphyritic, poikilitic, and occasionally graphic textures. The primary mineral assemblage of the pluton comprises plagioclase, orthoclase, quartz, amphibole, biotite, and Fe-Ti oxides. Accessory minerals include zircon, apatite, sphene, and allanite. Plagioclases are labradorite to oligoclase (An26 to An66) in composition. K-feldspar minerals exhibited an orthoclase composition (Or80 to Or97). All amphiboles belong to the calcic amphibole field and exhibit a magnesio-hornblende (Mg# 0.63-0.73) composition. Biotites crystallized as solidified melt products with compositions between annite and phlogopite endmembers, plotting close to the magnesium-rich (Mg#=0.52-0.58) end of the phlogopite solid solution series. Calculated crystallization temperatures derived from amphibole and biotite data range from 712°C to 824°C. Pressure estimations calculated using amphibole-plagioclase, amphibole and biotite suggest a range of 0.04 to 2.06 kbar. Oxygen fugacity (fO2) values calculated using amphibole and biotite fall between -12.5 and -16.1. Amphibole-based water content estimations indicate a range of 3.7% to 5.7% for the pluton. Biotite compositions within the studied Kazıkbeli pluton rocks exhibit characteristics suggestive of a potential mantle origin. Geobarometric calculations based on mineral chemistry data

*Received Date: 26.01.2024* with geological and petrographic features indicate the emplacement of the Kazıkbeli Pluton at relati-*Accepted Date: 07.08.2024* vely shallow depths within the crust (~1 to 8 km).

#### 1. Introduction

The Eastern Pontides Orogenic Belt (EPOB) constitutes a critical area for examining plutonic and volcanic lithologies (Okay and Şahintürk, 1997; Kaygusuz and Şahin, 2016; Özdamar, 2016; Temizel et al., 2019; Gücer and Sarı, 2021; Arslan et al., 2022;

Kaygusuz et al., 2022; Oğuz-Saka et al., 2023; Sar et al., 2023; Revan et al., 2023; Rezeau et al., 2023). The EPOB exhibits a rich tapestry of plutonic bodies, with emplacement ages spanning from the Carboniferous to the Eocene (Figure 1). Notably, these plutons display a compositional spectrum ranging from gabbro to granite (Figure 1). Further analysis reveals four

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distinct intrusive episodes within the EPOB: Paleozoic (Carboniferous), Jurassic, Late Cretaceous, and Eocene (Figure 1). Paleozoic plutons were intruded into metamorphic rocks (Yılmaz, 1972; Coğulu, 1975; Topuz et al., 2010; Dokuz, 2011; Kaygusuz et al., 2012, 2016; Ustaömer et al., 2013; Gücer and Sarı, 2021), Jurassic plutons were emplaced into pre-Jurassic basement rocks (Eyuboğlu et al., 2016; Karslı et al., 2017; Dokuz and Sünnetçi, 2019; Aydınçakır et al., 2020, 2023), Late Cretaceous plutons cut across subduction-related volcanic and/or volcaniclastic rocks (Kaygusuz, 2000; Boztuğ et al., 2006; İlbeyli, 2008; Kaygusuz et al., 2008, 2021; Liu et al., 2018; Eyuboğlu et al., 2019; Temizel et al., 2019, 2022; Yücel et al., 2023), and Eocene plutons intrude all previous units (Topuz et al., 2005, 2011; Karslı et al., 2007, 2011; Kaygusuz and Öztürk, 2015; Eyuboğlu et al., 2017, 2018; Kaygusuz et al., 2017, 2018, 2020; Özdamar et al., 2017; Sipahi et al., 2017; Temizel et al., 2018, 2020; Vural and Kaygusuz, 2021; Aydıncakır et al., 2022).

The chemical composition and evolution of the melt, as reflected by the mineral assemblage and the geochemical signature of the constituent minerals, are intrinsically linked in igneous rocks (Abbott, 1985). Variations in temperature (T), pressure (P), and oxygen fugacity (fO<sub>2</sub>) are recognized as the principal factors controlling mineral formation and stability within magma. Determining these intensive variables (T, P, fO<sub>2</sub>) offers valuable insights into the crystallization processes that govern magma evolution. Thermodynamic principles allow us to predict the mineralogical assemblage that crystallizes under specific conditions. While previous studies have addressed the emplacement conditions of some Eastern Pontide plutons (e.g., Karslı et al., 2007; Eyuboğlu et al., 2017; Kaygusuz et al., 2018, 2020; Temizel et al., 2018; Aydınçakır et al., 2020), the Kazıkbeli pluton remained unexamined in this context. This investigation, therefore, presents the first attempt to elucidate the pressure-temperature history of the Kazıkbeli pluton, offering new constraints on its crystallization environment.

Prior investigations of the Kazıkbeli Pluton have been confined to broad geological surveys (Güven, 1993). In-depth analyses were only recently undertaken by the lead author of this paper as part of their doctoral dissertation (Güloğlu, 2022). The present study delves into a specific aspect of Güloğlu's (2022) comprehensive doctoral research.

This investigation focuses on the petrographic and mineral chemical characteristics of the Middle Eocene Kazıkbeli Pluton exposed within the Kürtün region, Gümüşhane Province (Figure 2). By employing mineral chemistry data, the study aims to decipher the thermobarometric conditions (pressure-temperature) that prevailed during the pluton's crystallization. The findings will contribute to the existing body of knowledge regarding the emplacement conditions of the pluton.

# 2. Geological Position and Field Characteristics of the Kazıkbeli Pluton

The geologic record within the study area unveils a sequence of sedimentary and volcanic units (Figure 2). The oldest exposed unit is the Liassic-Doggerian aged Hamurkesen Formation (Ağar, 1977), composed primarily of basaltic and andesitic lavas with minor pyroclastics. Outcrops of this formation are limited to a small area north of the study region. Conformably overlying the Hamurkesen Formation is the widespread Late Cretaceous aged Catak Formation (Güven, 1993). This formation dominates the study area and is characterized by andesites at lower levels, transitioning upwards to tuff-dominated sequences with minor andesites and intercalations of limestones and tuffites. The Late Cretaceous stratigraphy continues with the Kızılkaya Formation (Güven, 1993), featuring widespread exposures of dacites and pyroclastics, overlying the Catak Formation. Another Late Cretaceous unit, the Cağlayan Formation (Güven, 1993), conformably overlies the aforementioned formations but with limited exposure in the study area's southwestern part. This formation is composed of basalt-andesite and associated pyroclastics. Unconformably above the Late Cretaceous formations lies the Eocene-aged Kabaköy Formation (Güven, 1993). This formation exhibits a distinct sequence, transitioning from nummulitic sandy limestone and tuffites through agglomerates to hornblende andesite and pyroclastics at upper levels. The Eocene-aged plutons, including Kazıkbeli Pluton intrudes all previously described formations. Quaternary alluviums represent the youngest geologic units within the study area (Figure 2).

The Kazıkbeli Pluton exhibits an elongated, elliptical shape with a northeast-southwest trending long axis (Figure 2). It covers an approximate area of

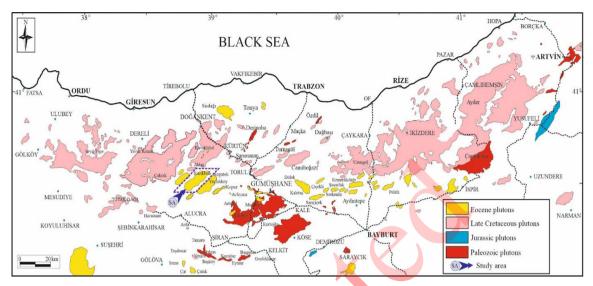


Figure 1- The location of the studied Kazıkbeli Pluton and the established distribution of plutonic rocks within the Eastern Pontides (modified after Güven, 1993 and Güloğlu, 2022).

46 km<sup>2</sup>, with dimensions ranging from 5-7 km in width and 8-10 km in length. The pluton displays intrusive contacts with a variety of formations spanning the Lower Jurassic to Eocene eras. These formations include the Hamurkesen Formation (Lower Jurassic), Çatak Formation (Late Cretaceous), Kızılkaya Formation (Late Cretaceous), Çağlayan Formation (Late Cretaceous), and Kabaköy Formation (Eocene). The contact zones are characterized by significant alteration, evidenced by localized epidotization and chloritization. Faulted contact zones exhibit further evidence of strike-slip and normal fault movement, with brecciated structures and crush zones prevalent in areas affected by normal faulting.

The Kazıkbeli Pluton is characterized by distinct cooling fractures, particularly around Kazıkbeli Plateau, Beytarla, Çurufdibi Plateau, Davunlu Plateau, Dikme Plateau, and Gavurluk Plateau (Figures 3a and b). These areas exhibit well-developed, sporadic crack systems and well-blocked rock structures (Figures 3a and b). In contrast, plutonic rocks near the margins display more intensive fracturing and cracking, lacking a well-blocked structure. Overall, the Kazıkbeli Pluton presents a generally robust appearance with minimal weathering. However, in weathered zones, the rocks become friable, acquiring a sandy and earthy texture (Figures 3c and d).

#### 3. Material and Method

Within the scope of the research, a total of

50 rock samples were collected during fieldwork, and thin sections of these samples were prepared to determine detailed petrographic features under a polarizing microscope. In addition, modal analyses were conducted for 39 of these samples. In the modal analyses, counts were made for sections with grain sizes between approximately 0.5 and 0.8 mm at 1000 to 1100 points, and for sections with grain sizes between 1.0 and 1.5 mm at 1500 to 1600 points. Counting errors were calculated for the counted samples using error formulas and counts for four samples were repeated. The mineral chemistry (microprobe) analyses of two rock samples (gabbroic diorite and granodiorite) from the studied plutonic rock were performed at the Geological and Mineral Research Laboratory of New Mexico Tech University in the United States. The analyses of previously prepared polished section samples were conducted using a CAMECA SX 100 Electron Microprobe under conditions of 15 kV voltage and 20 nA current. The beam diameter used in the analyses was 10 µm, and counting times were set at 10 seconds for Al, Si, Ti, Ca, Mn, Fe, Mg, K, and Na elements. Electron microprobe analyses employed a 1 µm focused beam for pyroxene, amphibole, epidote, zircon and Fe-Ti oxide. To minimize Na (sodium) loss during analyses of mica, feldspar, and chlorite, a slightly defocused 10 µm beam was utilized. The standards used for measurements were diopside, kaersutite, albite, orthoclase, biotite, magnetite, and anorthite (UCB) (Nielsen and Sigurdsson, 1981).

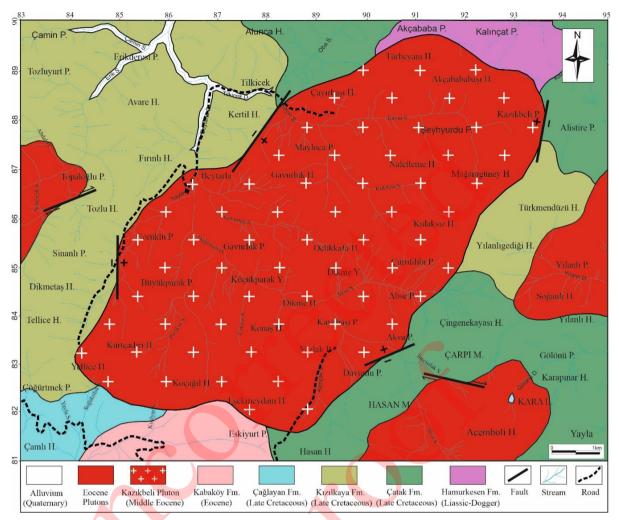


Figure 2- Geological map of the study area (modified from Güloğlu, 2022).



Figure 3- a-b) Cracking systems in the rocks of the Kazıkbeli Pluton, c-d) Arena formation observed in the rocks of the Kazıkbeli Pluton.

#### 4. Results

#### 4.1. Petrography of the Kazıkbeli Pluton

The results of the modal analysis for the investigated plutonic rocks are presented in Table 1. The modal mineralogy is further visualized on a QAP diagram (Figure 4) and a zoning map of the pluton based on modal compositions (Figure 5).

Modal analysis data for the Kazıkbeli Pluton rock samples, plotted on the QAP diagram (Streckeisen, 1976; Figure 4), reveals a compositional spectrum encompassing diorite. quartz diorite, tonalite, granodiorite, and monzogranite, Notably, monzogranites, the least abundant rock type, are primarily concentrated in the central regions of the pluton (Figure 5). Granodiorites exhibit the most widespread distribution, commonly occurring around the periphery of the monzogranite zone (Figure 5). Tonalites, quartz diorites, and diorites are less prevalent compared to granodiorites and are typically situated at the pluton's margins (Figure 5). This spatial distribution suggests a compositional zonation within the Kazıkbeli Pluton.

Microscopic examination reveals that the Kazıkbeli Pluton rocks exhibit a range of textures, including fine- to medium-grained, porphyritic, occasionally graphic, poikilitic, and perthitic (Figures 6a-d). The primary mineral assemblage of the pluton comprises plagioclase, quartz, orthoclase, amphibole, biotite, and opaque minerals. Accessory minerals include apatite and zircon. Secondary minerals identified in the rocks are clay minerals, sericite, chlorite, and calcite.

Table 1- Modal mineralogy of the Kazıkbeli Pluton, summarized by minimum, maximum, and average values obtained from modal analyses of the rock samples.

Rock type	Plagioclase	Quartz	Orthoclase	Amphibole	Biotite	Opaque minerals
<b>Diorite</b>	Diorite					
min	63.80	1.80	3.00	11.20	0.80	2.30
max	80.40	3.70	5.20	21.40	2.20	6.10
avg	72.48	2.60	4.18	15.16	1.68	4.14
Quartz diorite						
min	56.50	6.50	3.30	13.40	0.60	4.30
max	64.00	13.95	5.82	20.50	5.70	4.80
avg	61.09	9.56	4.23	16.42	4.64	4.52
<u>Tonalite</u>						
min	37.10	19.72	7.40	1.90	0.90	1.35
max	62.10	30.90	21.02	14.30	8.70	3.90
avg	48.14	26.11	13.51	6.04	4.22	2.25
<u>Granodiorite</u>						
min	60.30	18.20	3.20	5.30	1.00	2.10
max	61.90	24.10	5.60	12.40	2.20	2.90
avg	60.90	21.27	4.43	9.63	1.53	2.67
<b>Monzogranite</b>						
min	26.04	26.10	21.80	1.30	1.30	0.65
max	39.10	37.00	30.00	7.70	7.39	3.00
avg	33.69	32.07	26.12	3.53	4.02	1.98

min: minimum values, max: maximum values, avg: average values

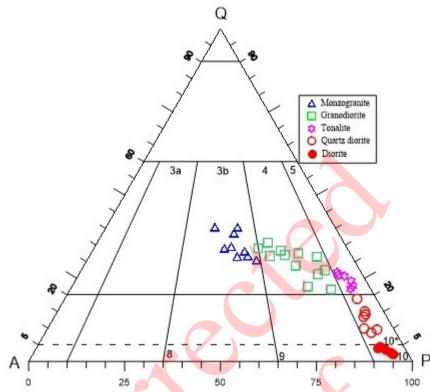


Figure 4- The QAP classification diagram (Streckeisen, 1976) illustrates the modal compositions of rock samples from the Kazıkbeli Pluton (3a: syenogranite, 3b: monzogranite, 4: granodiorite, 5: tonalite, 8: monzonite, 9: monzodiorite/gabbro, 10: diorite/gabbro, 10\*: quartz diorite/gabbro).

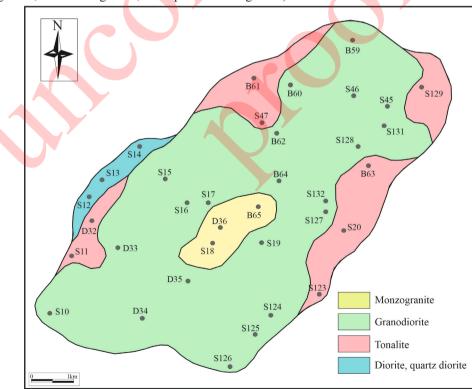


Figure 5- Locations and zoning map based on modal compositions of rock samples from the Kazıkbeli Pluton (S, B, D: sample locations).

Plagioclase is the most abundant mineral, present in all samples (26.04-80.40%; Table 1) as idiomorphic to subhedral tabular prismatic crystals (Figures 6a-d). Albite twins are commonly observed, with occasional occurrences of polysynthetic (albitepericline) twins. Oscillatory zoning is present in some larger crystals (Figure 6d). K-feldspar (orthoclase) is anhedral to subhedral crystals and is found in rocks at a rate of 3.0-30.0% (Table 1). Karlsbad twinning is observed in some minerals, while others exhibit perthitic structures. Notably, some large orthoclase crystals enclose smaller plagioclase crystals, creating a poikilitic texture (Figure 6c). Additionally, in some instances, orthoclase fills the space between quartz and other minerals (Figure 6b). Quartz is anhedral to subhedral crystals of varying sizes. Undulatory extinction is observed in some sections, indicating a fractured and cracked structure (Figure 6c). The abundance of quartz varies between 1.8% and 37% (Table 1). Amphibole is idiomorphic to subhedral tabular prismatic crystals. Larger crystals may contain inclusions of opaque minerals and plagioclase (Figures 6a and 6b). Alteration in some rock samples resulted in the formation of chlorite and calcite. The abundance varies between 1.3% and 21.4% (Table 1). Biotite is idiomorphic to subhedral acicular prismatic crystals, and is found at a rate of 0.6-8.7% (Table 1). Color pleochroism is evident in some sections, with partial alteration of biotite to chlorite observed in association with feldspar (Figure 6c). Apatite is typically occurs as needle-like inclusions within quartz and feldspar. Zircon is observed in almost all rocks as small, euhedral prismatic crystals. Opaque Minerals are present as euhedral and subhedral crystals, and are found at a rate of 0.7-6.1% of the studied rocks (Table 1).

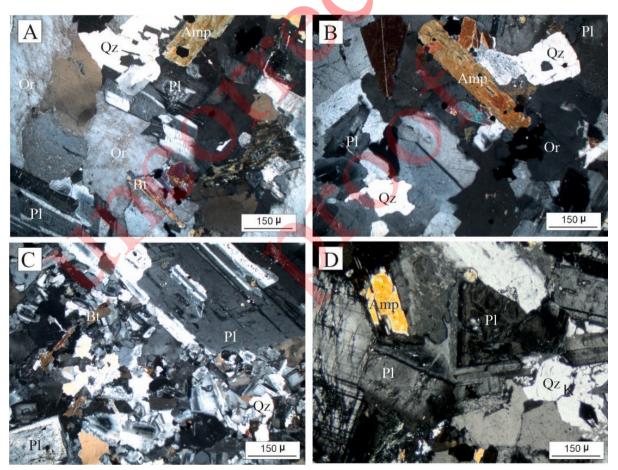


Figure 6- Microscopic features observed in the rocks of the studied pluton, a) Fine- to medium-grained texture with large orthoclase crystals, b) Granular texture, c) Poikilitic texture, characterized by small plagioclase crystals enclosed within larger plagioclase phenocrysts, d) Zoned plagioclase crystal (Cross Nicol, Pl: plagioclase, Or: orthoclase, Qz: quartz, Amp: amphibole Bt: biotite).

#### 4.2. Mineral Chemistry

#### 4.2.1. Plagioclase

Microprobe analysis of plagioclase minerals in the studied Kazıkbeli Pluton reveals compositional

Table 2- Microprobe analysis values of plagioclase.

variations between  $An_{26}$  and  $An_{66}$  (Table 2; Figure 7). Plagioclases within both mafic (gabbroic diorite) and felsic (granodiorite) lithologies exhibit SiO<sub>2</sub> contents ranging from 51 to 61 wt%,  $Al_2O_3$  from 24 to 30 wt%, FeO<sup>T</sup> from 0.2 to 0.5 wt%, and K<sub>2</sub>O from 0.2 to 0.5 wt% (Table 2).

	S 13-01	c diorite S 13-02	S 13-03	IS 13-04	IS 13-05	IS 13-06	5IS 13-07	IS 13-08	3 IS 13-11
Sample (r-c)	(r)	(r)	(c)	(c)	(c)	(c)	$\frac{(c)}{(c)}$	(c)	$\frac{(r)}{(r)}$
SiO <sub>2</sub>	55.39	54.74	51.91	54.69	51.26	54.08	53.90	53.03	50.96
$Al_2O_3$	27.84	27.68	30.02	27.87	30.05	28.02	28.25	28.59	30.34
FeOT	0.35	0.33	0.44	0.48	0.37	0.36	0.38	0.51	0.35
CaO	9.92	10.45	12.71	10.17	13.02	10.81	10.90	11.26	13.33
Na <sub>2</sub> O	5.97	5.44	4.02	5.67	3.89	5.12	4.91	4.62	3.76
K <sub>2</sub> O	0.29	0.32	0.23	0.31	0.19	0.34	0.31	0.29	0.18
BaO	0.06	0.06	0.06	0.07	0.04	0.05	0.05	0.06	0.03
SrO	0.10	0.17	0.09	0.14	0.12	0.09	0.15	0.10	0.17
Total	99.92		99.48	99.40	98.94	98.87	98.85	98.46	99.12
Si	2.50		2.37	2.49	2.36	2.47	2.47	2.44	2.34
Al	1.48	1.49	1.62	1.50	1.63	1.51	1.52	1.55	1.64
Fe <sup>(ii)</sup>	0.01	0.01	0.02	0.02	0.01	0.01	0.01	0.02	0.01
Ca	0.48	0.51	0.62	0.50	0.64	0.53	0.53	0.56	0.66
Na	0.52	0.48	0.36	0.50	0.35	0.45	0.44	0.41	0.34
K	0.02	0.02	0.01	0.02	0.01	0.02	0.02	0.02	0.01
An	47.08		62.74	48.90	64.18	52.78	54.08	56.40	65.51
Ab	51.28		35.91	49.33	34.70	45 <mark>.</mark> 24	44.09	41.87	33.44
Or FeO <sup>T</sup> has b	1.64	1.84	1.35	1.77	1.12	1.98	1.83	1.73	1.05
Rock type	Crano	diorito							
Sample 7	S 19-0	<b>diorite</b> 1  S 19-02	2  S 19-03	3 <u> S 19-07</u>	IS 19-08	<u> S 19-11</u>	<u>S 19-12</u>	S 19	-15
Sample <sup>*</sup> (r-c)		0diorite 1  S 19-02  (c)  55.39	2  S 19-03 (c) 55.11	3 S 19-07 (c) 59.40	S 19-08 (r) 60.61	S 19-11 (c) 57.06	S 19-12 (r) 61.16	S 19 (r) 59.7	
Sample (r-c) SiO <sub>2</sub>	S 19-0 (r)	1 S 19-02 (c)	(c)	(c)	$(\mathbf{r})$	(c)	(r)	(r)	8
Sample (r-c) SiO <sub>2</sub> $AI_2O_3$ FeO <sup>T</sup>	S 19-0 (r) 56.03 26.97 0.47	1 <u>S 19-02</u> (c) 55.39	(c) 55.11	(c) 59.40	(r) 60.61 24.51 0.24	(c) 57.06 26.36 0.33	(r) 61.16 23.86 0.26	(r) 59.7	8 2
Sample (r-c) SiO <sub>2</sub> $AI_2O_3$ FeO <sup>T</sup>	\$ 19-0 (r) 56.03 26.97 0.47 9.28	1 S 19-02 (c) 55.39 27.54 0.48 9.63	(c) 55.11 27.42 0.49 9.52	(c) 59.40 24.59 0.30 6.29	(r) 60.61 24.51 0.24 6.04	(c) 57.06 26.36 0.33 8.39	(r) 61.16 23.86 0.26 5.40	(r) 59.7 25.0	8
$\frac{\text{Sample}^{T}}{\text{(r-c)}}$ SiO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub> FeO <sup>T</sup> CaO	S 19-0 (r) 56.03 26.97 0.47	1 <u>S 19-02</u> (c) 55.39 27.54 0.48	(c) 55.11 27.42 0.49	(c) 59.40 24.59 0.30	(r) 60.61 24.51 0.24	(c) 57.06 26.36 0.33	(r) 61.16 23.86 0.26	(r) 59.7 25.0 0.32	8
$\frac{\text{Sample}^{T}}{(r-c)}$ SiO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub> FeO <sup>T</sup> CaO Na <sub>2</sub> O	\$ 19-0 (r) 56.03 26.97 0.47 9.28	1 S 19-02 (c) 55.39 27.54 0.48 9.63	(c) 55.11 27.42 0.49 9.52	(c) 59.40 24.59 0.30 6.29	(r) 60.61 24.51 0.24 6.04	(c) 57.06 26.36 0.33 8.39	(r) 61.16 23.86 0.26 5.40	(r) 59.7 25.0 0.32 6.83	8
Sample (r-c) SIO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub> FeO <sup>T</sup> CaO Na <sub>2</sub> O K <sub>2</sub> O BaO	S 19-0 (r)           56.03           26.97           0.47           9.28           6.19           0.39           0.06	1 S 19-02 (c) 55.39 27.54 0.48 9.63 5.97 0.38 0.05	(c) 55.11 27.42 0.49 9.52 5.98 0.38 0.08	(c) 59.40 24.59 0.30 6.29 8.02 0.44 0.03	(r) 60.61 24.51 0.24 6.04 8.26 0.46 0.46 0.00	(c) 57.06 26.36 0.33 8.39 6.85 0.34 0.06	(r) 61.16 23.86 0.26 5.40 8.34 0.37 0.00	(r) 59.7 25.0 0.32 6.83 7.44 0.47 0.01	8
Sample $(r-c)$ $SiO_2$ $AI_2O_3$ $FeO^T$ $CaO$ $Na_2O$ $K_2O$ $BaO$ $SrO$	S 19-0 (r) 56.03 26.97 0.47 9.28 6.19 0.39 0.06 0.10	1         S         19-02           (c)         55.39           27.54         0.48           9.63         5.97           0.38         0.05           0.13         0.13	(c) 55.11 27.42 0.49 9.52 5.98 0.38 0.08 0.12	(c) 59.40 24.59 0.30 6.29 8.02 0.44 0.03 0.08	(r) 60.61 24.51 0.24 6.04 8.26 0.46 0.00 0.04	(c) 57.06 26.36 0.33 8.39 6.85 0.34 0.06 0.04	(r) 61.16 23.86 0.26 5.40 8.34 0.37 0.00 0.06	(r) 59.7 25.0 0.32 6.83 7.44 0.47 0.01 0.10	8
Rock typeSample $(r-c)$ SiO2 $AI_2O3$ FeO1CaONa2OK2OBaOSrOTotalCi	S 19-0 (r)           56.03           26.97           0.47           9.28           6.19           0.39           0.06           0.10           99.49	1         S         19-02           (c)         55.39           27.54         0.48           9.63         5.97           0.38         0.05           0.13         99.57	(c) 55.11 27.42 0.49 9.52 5.98 0.38 0.08 0.12 99.10	(c) 59.40 24.59 0.30 6.29 8.02 0.44 0.03 0.08 99.15	(r) 60.61 24.51 0.24 6.04 8.26 0.46 0.00 0.04 100.16	(c) 57.06 26.36 0.33 8.39 6.85 0.34 0.06 0.04 99.43	(r) 61.16 23.86 0.26 5.40 8.34 0.37 0.00 0.06 99.45	(r) 59.7 25.0 0.32 6.83 7.44 0.47 0.01 0.10 99.9	8
Sample (r-c) SiO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub> FeO <sup>T</sup> CaO Na <sub>2</sub> O K <sub>2</sub> O BaO SrO Total Si	S 19-0 (r)           56.03           26.97           0.47           9.28           6.19           0.39           0.06           0.10           99.49           2.54	1         S         19-02           (c)         55.39           27.54         0.48           9.63         5.97           0.38         0.05           0.13         99.57           2.51         1	(c) 55.11 27.42 0.49 9.52 5.98 0.38 0.38 0.08 0.12 99.10 2.51	(c) 59.40 24.59 0.30 6.29 8.02 0.44 0.03 0.08 99.15 2.68	(r) 60.61 24.51 0.24 6.04 8.26 0.46 0.00 0.04 100.16 2.70	(c) 57.06 26.36 0.33 8.39 6.85 0.34 0.06 0.04 99.43 2.58	(r) 61.16 23.86 0.26 5.40 8.34 0.37 0.00 0.06 99.45 2.74	(r) 59.7 25.0 0.32 6.83 7.44 0.47 0.01 0.10 99.9 2.67	8 2 7
Sample $(r-c)$ SiO2 $AI_2O3$ FeO1CaONa2OK2OBaOSrOTotalSiAl	S 19-0 (r)           56.03           26.97           0.47           9.28           6.19           0.39           0.06           0.10           99.49           2.54           1.44	1         S         19-02           (c)         55.39           27.54         0.48           9.63         5.97           0.38         0.05           0.13         99.57           2.51         1.47	(c) 55.11 27.42 0.49 9.52 5.98 0.38 0.38 0.08 0.12 99.10 2.51 1.47	(c) 59.40 24.59 0.30 6.29 8.02 0.44 0.03 0.08 99.15 2.68 1.31	(r) 60.61 24.51 0.24 6.04 8.26 0.46 0.46 0.00 0.04 100.16 2.70 1.29	(c) 57.06 26.36 0.33 8.39 6.85 0.34 0.06 0.04 99.43 2.58 1.40	(r) 61.16 23.86 0.26 5.40 8.34 0.37 0.00 0.06 99.45 2.74 1.26	(r) 59.7 25.0 0.32 6.83 7.44 0.47 0.01 0.10 99.9 2.67 1.32	8 2 7
Sample $(r-c)$ $SiO_2$ $Al_2O_3$ $FeO^T$ $CaO$ $Na_2O$ $K_2O$ $BaO$ $SrO$ TotalSi $Al$ $Fe^{(ii)}$	S 19-0 (r)           56.03           26.97           0.47           9.28           6.19           0.39           0.06           0.10           99.49           2.54           1.44           0.02	1         S         19-02           (c)         55.39           27.54         0.48           9.63         5.97           0.38         0.05           0.13         99.57           2.51         1.47           0.02         1.47	(c) 55.11 27.42 0.49 9.52 5.98 0.38 0.08 0.12 99.10 2.51 1.47 0.02	(c) 59.40 24.59 0.30 6.29 8.02 0.44 0.03 0.08 99.15 2.68 1.31 0.01	(r) 60.61 24.51 0.24 6.04 8.26 0.46 0.00 0.04 100.16 2.70 1.29 0.01	(c) 57.06 26.36 0.33 8.39 6.85 0.34 0.06 0.04 99.43 2.58 1.40 0.01	(r) 61.16 23.86 0.26 5.40 8.34 0.37 0.00 0.06 99.45 2.74 1.26 0.01	(r) 59.7 25.0 0.32 6.83 7.44 0.47 0.01 0.10 99.9 2.67 1.32 0.01	8 2 7 7
Sample $(r-c)$ $SiO_2$ $AI_2O_3$ $FeO^T$ $CaO$ $Na_2O$ $K_2O$ $BaO$ $SrO$ $Total$ $Si$ $AI$ $Fe^{(ii)}$ $Ca$	S 19-0 (r)           56.03           26.97           0.47           9.28           6.19           0.39           0.06           0.10           99.49           2.54           1.44           0.02           0.45	1         S         19-02           (c)         55.39           27.54         0.48           9.63         5.97           0.38         0.05           0.13         99.57           2.51         1.47           0.02         0.47	(c) 55.11 27.42 0.49 9.52 5.98 0.38 0.08 0.12 99.10 2.51 1.47 0.02 0.47	(c) 59.40 24.59 0.30 6.29 8.02 0.44 0.03 0.08 99.15 2.68 1.31 0.01 0.30	(r) 60.61 24.51 0.24 6.04 8.26 0.46 0.00 0.04 100.16 2.70 1.29 0.01 0.29	(c) 57.06 26.36 0.33 8.39 6.85 0.34 0.06 0.04 99.43 2.58 1.40 0.01 0.01 0.41	(r) 61.16 23.86 0.26 5.40 8.34 0.37 0.00 0.06 99.45 2.74 1.26 0.01 0.26	(r) 59.7 25.0 0.32 6.83 7.44 0.47 0.01 0.10 99.9 2.67 1.32 0.01 0.33	8 2 7 7
Sample $(r-c)$ $SiO_2$ $AI_2O_3$ $FeO^T$ $CaO$ $Na_2O$ $K_2O$ $BaO$ $SrO$ $Total$ $Si$ $AI$ $Fe^{(ii)}$ $Ca$ $Na$	S 19-0 (r)           56.03           26.97           0.47           9.28           6.19           0.39           0.06           0.10           99.49           2.54           1.44           0.02           0.45           0.54	1         S         19-02           (c)         55.39           27.54         0.48           9.63         5.97           0.38         0.05           0.13         99.57           2.51         1.47           0.02         0.47           0.53         0.53	(c)           55.11           27.42           0.49           9.52           5.98           0.38           0.08           0.12           99.10           2.51           1.47           0.02           0.47	(c)           59.40           24.59           0.30           6.29           8.02           0.44           0.03           0.08           99.15           2.68           1.31           0.01           0.30           0.70	(r) 60.61 24.51 0.24 6.04 8.26 0.46 0.00 0.04 100.16 2.70 1.29 0.01 0.29 0.71	(c) 57.06 26.36 0.33 8.39 6.85 0.34 0.06 0.04 99.43 2.58 1.40 0.01 0.41 0.60	(r)           61.16           23.86           0.26           5.40           8.34           0.37           0.00           0.06           99.45           2.74           1.26           0.01           0.26           0.72	(r) 59.7 25.0 0.32 6.83 7.44 0.47 0.01 0.10 99.9 2.67 1.32 0.01 0.33 0.64	8 2 7 7
Sample $(r-c)$ $SiO_2$ $AI_2O_3$ $FeO^T$ $CaO$ $Na_2O$ $K_2O$ $BaO$ $SrO$ $Total$ $Si$ $AI$ $Fe^{(ii)}$ $Ca$ $Na$ $K$	S 19-0 (r)           56.03           26.97           0.47           9.28           6.19           0.39           0.06           0.10           99.49           2.54           1.44           0.02           0.45           0.54           0.02	1         S         19-02           (c)         55.39           27.54         0.48           9.63         5.97           0.38         0.05           0.13         99.57           2.51         1.47           0.02         0.47	(c)           55.11           27.42           0.49           9.52           5.98           0.38           0.08           0.12           99.10           2.51           1.47           0.02           0.47           0.53           0.02	(c) 59.40 24.59 0.30 6.29 8.02 0.44 0.03 0.08 99.15 2.68 1.31 0.01 0.30 0.70 0.03	(r) 60.61 24.51 0.24 6.04 8.26 0.46 0.00 0.04 100.16 2.70 1.29 0.01 0.29 0.71 0.03	(c) 57.06 26.36 0.33 8.39 6.85 0.34 0.06 0.04 99.43 2.58 1.40 0.01 0.41 0.60 0.02	(r)           61.16           23.86           0.26           5.40           8.34           0.37           0.00           0.06           99.45           2.74           1.26           0.01           0.26           0.72           0.02	(r) 59.7 25.0 0.32 6.83 7.44 0.47 0.01 0.10 99.9 2.67 1.32 0.01 0.33 0.64 0.03	8 2 7 7
Sample $(r-c)$ $SiO_2$ $AI_2O_3$ $FeO^T$ $CaO$ $Na_2O$ $K_2O$ $BaO$ $SrO$ $Total$ $Si$ $AI$ $Fe^{(ii)}$ $Ca$ $Na$	S 19-0 (r)           56.03           26.97           0.47           9.28           6.19           0.39           0.06           0.10           99.49           2.54           1.44           0.02           0.45           0.54	1         S         19-02           (c)         55.39           27.54         0.48           9.63         5.97           0.38         0.05           0.13         99.57           2.51         1.47           0.02         0.47           0.53         0.53	(c)           55.11           27.42           0.49           9.52           5.98           0.38           0.08           0.12           99.10           2.51           1.47           0.02           0.47	(c)           59.40           24.59           0.30           6.29           8.02           0.44           0.03           0.08           99.15           2.68           1.31           0.01           0.30           0.70	(r) 60.61 24.51 0.24 6.04 8.26 0.46 0.00 0.04 100.16 2.70 1.29 0.01 0.29 0.71	(c) 57.06 26.36 0.33 8.39 6.85 0.34 0.06 0.04 99.43 2.58 1.40 0.01 0.41 0.60	(r)           61.16           23.86           0.26           5.40           8.34           0.37           0.00           0.06           99.45           2.74           1.26           0.01           0.26           0.72	(r) 59.7 25.0 0.32 6.83 7.44 0.47 0.01 0.10 99.9 2.67 1.32 0.01 0.33 0.64	8 2 7 7 5

Table 2-

Plagioclases within the gabbroic diorites of the Kazıkbeli Pluton display oscillatory zoning, characterized by fluctuations in An content between  $An_{47}$  and  $An_{64}$  (Table 2).

#### 4.2.2. K-Feldspars

Examination of Eocene-aged rocks revealed the presence of K-feldspar minerals throughout most

samples. Chemical analyses of these K-feldspar minerals (Table 3) documented variations in SiO<sub>2</sub> (63-65 wt%), Al<sub>2</sub>O<sub>3</sub> (19-20 wt%), and BaO (0.03-0.72 wt%) (Table 3; Figure 8). Notably, all K-feldspar minerals exhibited an orthoclase composition (Figure 9) with no evidence of chemical zoning within the crystals. The overall rock compositions ranged from  $Or_{80}$  to  $Or_{97}$  (Table 3; Figure 8).

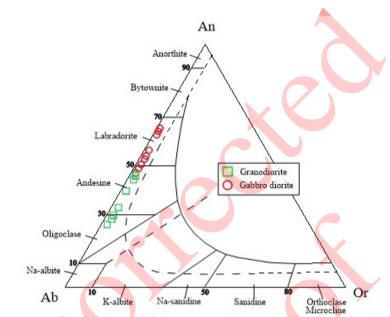


Figure 7- An-Ab-Or ternary diagram showing felspar compositions (Smith and Brown, 1988) in the studied pluton.

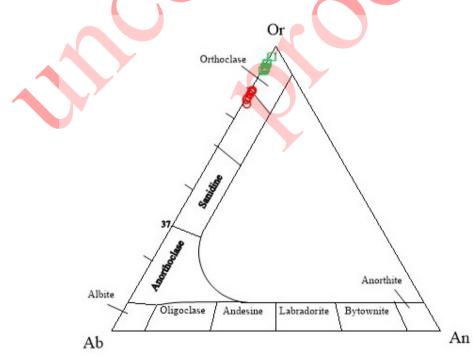


Figure 8- Classification diagram for K-feldspars using the An-Ab-Or diagram (from Deer et al., 1992).

Rock type	Gabbroic diorite								
Sample	S 13-1	S 13-2	S 13-3	S 13-4	S 13-5	S 13-6	S 13-7	S 13-8	
(r-c)	(r)	(c)	(c)	(c)	(c)	(c)	(c)	(r)	
SiO <sub>2</sub>	64.33	64.65	64.42	64.32	64.27	64.36	64.80	64.86	
Al <sub>2</sub> O <sub>3</sub>	18.91	19.12	19.04	19.15	19.14	19.52	19.02	19.08	
FeO <sup>T</sup>	0.30	0.12	0.10	0.10	0.26	0.16	0.09	0.15	
CaO	0.11	0.13	0.08	0.11	0.09	0.27	0.10	0.08	
Na <sub>2</sub> O	1.78	1.70	2.01	1.83	2.07	2.11	1.69	1.69	
K <sub>2</sub> O	14.58	14.70	14.09	14.13	13.89	13.43	14.24	14.37	
BaO	0.03	0.07	0.21	0.28	0.26	0.72	0.12	0.09	
SrO	0.03	0.03	0.05	0.09	0.03	0.08	0.07	0.07	
Total	100.07	100.52	100.00	100.01	100.01	100.65	100.13	100.39	
Si	2.97	2.97	2.97	2.96	2.96	2.95	2.98	2.97	
Al	1.03	1.03	1.03	1.04	1.04	1.05	1.03	1.03	
Fe <sup>(ii)</sup>	0.01	0.00	0.00	0.00	0.01	0.01	0.00	0.01	
Ca	0.01	0.01	0.00	0.01	0.00	0.01	0.00	0.00	
Na	0.16	0.15	0.18	0.16	0.18	0.19	0.15	0.15	
K	0.86	0.86	0.83	0.83	0.82	0.78	0.83	0.84	
Ba	0	0	0	0.01	0	0.01	0	0	
An	0.53	0.63	0.39	0.54	0.44	1.34	0.50	0.40	
Ab	15.57	14.86	17.75	16.36	18.39	19.02	15.21	15.10	
Or	83.90	84.52	81.86	83.10	81.17	79.64	84.30	84.50	

Table 3- Microprobe analysis values of K-feldspars.

FeO<sup>T</sup> has been measured as total iron. The formula has been calculated based on 8 oxygen. r: rim. c: center. Table 3- Continued.

Rock type 💋	Granodi	orite						
Sample	S 19-1	S 19-2	S 19-3	<u>S 19-4</u>	<u>S 19-5</u>	<u>S 19-6</u>	S 19-7	<u>S 19-8</u>
(r-c)	(r)	(c)	(c)	(c)	(c)	(c)	(c)	(r)
SiO <sub>2</sub>	63.80	63.57	63.23	63.60	64.99	63.68	63.86	63.80
Al <sub>2</sub> O <sub>3</sub>	18.97	18.96	18.73	18.70	19.14	18.88	18.92	18.93
FeO <sup>T</sup>	0.09	0.04	0.13	0.21	0.03	0.04	0.10	0.01
CaO	0.09	0.11	0.07	0.10	0.06	0.05	0.13	0.12
Na <sub>2</sub> O	0.79	0.79	0.67	0.83	0.95	0.62	0.84	0.35
K <sub>2</sub> O	15.86	15.69	15.62	15.57	15.70	15.74	15.21	16.28
BaO	0.18	0.51	0.45	0.24	0.21	0.40	0.54	0.29
SrO	0.05	0.07	0.03	0.05	0.08	0.06	0.10	0.03
Total	99.83	99.74	98.93	99.30	101.16	99.47	99.70	99.81
Si	2.96	2.96	2.97	2.97	2.97	2.97	2.97	2.97
Al	1.04	1.04	1.04	1.03	1.03	1.04	1.04	1.04
Fe <sup>(ii)</sup>	0.00	0.00	0.01	0.01	0.00	0.00	0.00	0.00
Са	0.00	0.01	0.00	0.01	0.00	0.00	0.01	0.01
Na	0.07	0.07	0.06	0.08	0.08	0.06	0.08	0.03
K	0.94	0.93	0.93	0.93	0.92	0.94	0.90	0.97
Ba	0	0.01	0.01	0	0	0.01	0.01	0.01

#### 4.2.3. Biotite

Microprobe analyses of biotites (Table 4) reveal TiO<sub>2</sub> contents ranging from 4.1 to 4.7 wt%, Al<sub>2</sub>O<sub>3</sub> from 12 to 13 wt%, and MgO from 12 to 14 wt%. Mg# (Mg/ (Mg + Fe<sup>+2</sup>)) and Fe# (Fe<sup>+2</sup>/(Fe<sup>+2</sup> + Mg)) ratios range from 0.52 to 0.58 and 0.42 to 0.48, respectively (Table 4). These values indicate that biotites crystallized as solidified melt products with compositions between annite and phlogopite endmembers, plotting close to the magnesium-rich end of the phlogopite solid solution series (Figure 9a). Furthermore, microprobe analyses classify biotites within the C field of the FeO<sup>T</sup>-Al<sub>2</sub>O<sub>3</sub> diagram (Figure 9b), coexisting with pyroxenes and hornblendes based on the FeO<sup>T</sup> - Al<sub>2</sub>O<sub>3</sub>-MgO

triangular diagram (Figure 9c). Finally, the MgO-FeO+MnO-TiO<sub>2</sub>\*10 triangular diagram classifies these biotites as primary brown biotites (A field) (Figure 9d).

#### 4.2.4. Amphibole

Microprobe analyses of amphiboles from the studied plutonic rocks (Table 5) reveal  $Al_2O_3$  contents ranging from 4.6 to 6.2 wt%,  $SiO_2$  contents from 47.1 to 50.2 wt%, and Mg# values between 0.63 and 0.73. Classification diagrams indicate that all amphiboles belong to the calcic amphibole field based on the Na versus (Ca+Na) ratio (Figure 10a) and exhibit a magnesiohornblende composition according to Leake et al. (1997) (Figure 10b).

Rock type	Gabbro	oic diorite	<u> </u>	Granod	iorite					
Sample	S 13-1	S 13-2	S 13-3	S 19-1	S 19-2	S 19-3	S 19-4	S 19-5	S 19-6	S 19-7
(r-c)	(r)	(c)	(c)	(r)	(c)	(r)	(c)	(c)	(c)	(c)
SiO <sub>2</sub>	37.85	37.38	37.91	36.69	36.02	37.49	36.94	36.79	37.11	37.14
TiO <sub>2</sub>	4.59	4.64	4.69	4.12	4.14	4.30	4.11	4.43	4.68	4.61
Al <sub>2</sub> O <sub>2</sub>	12.40	12.47	12.71	13.04	12.67	12.20	12.60	12.52	12.60	12.75
FeO <sup>T</sup>	18.02	18.23	17.57	19.44	19.31	18.96	19.28	18.70	18.51	19.03
MnO	0.26	0.26	0.22	0.22	0.17	0.19	0.18	0.33	0.31	0.32
MgO	13.31	12.95	13.84	11.81	11.57	12.03	12.21	12.08	12.80	12.34
CaO	0.04	0.05	0.04	0.20	0.27	0.19	0.11	0.27	0.06	0.09
Na <sub>2</sub> O	0.12	0.13	0.14	0.15	0.16	0.15	0.16	0.14	0.14	0.15
K,O	9.50	9.44	9.50	8.67	9.11	8.73	9.10	9.02	9.32	9.28
CI	0.19	0.26	0.24	0.32	0.33	0.30	0.25	0.27	0.41	0.27
F	0.40	0.38	0.44	0.14	0.14	0.18	0.19	0.32	0.31	0.35
Total	96.68	96.19	97.30	94.80	93.89	94.72	95.13	94.87	96.25	96.33
Si	5.65	5.62	5.60	5.62	5.60	5.73	5.65	5.62	5.59	5.59
Ti	0.52	0.52	0.52	0.47	0,48	0.49	0.47	0.51	0.53	0.52
Al	2.18	2.21	2.21	2.36	2.32	2.20	2.27	2.26	2.23	2.26
Fe <sup>(ii)</sup>	2.25	2.29	2.17	2.49	2.51	2.42	2.46	2.39	2.33	2.40
Mn	0.03	0.03	0.03	0.03	0.02	0.02	0.02	0.04	0.04	0.04
Mg	2.96	2.90	3.05	2.70	2.68	2.74	2.78	2.75	2.87	2.77
Са	0.01	0.01	0.01	0.03	0.04	0.03	0.02	0.04	0.01	0.01
Na	0.03	0.04	0.04	0.04	0.05	0.04	0.05	0.04	0.04	0.04
К	1.81	1.81	1.79	1.69	1.81	1.70	1.77	1.76	1.79	1.78
Cl	0.05	0.07	0.06	0.08	0.09	0.08	0.06	0.07	0.10	0.07
F	0.19	0.18	0.21	0.07	0.07	0.09	0.09	0.15	0.15	0.17
Fe#	0.43	0.44	0.42	0.48	0.48	0.47	0.47	0.46	0.45	0.46

Table 4- The microprobe analysis results of biotites.

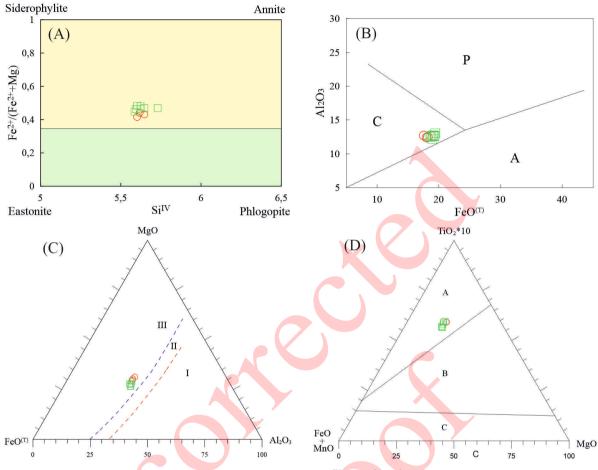


Figure 9- The composition of biotites in the studied rocks: a)  $Si^{IV} - (Fe^{+2}/Fe^{+2} + Mg)$  diagram (Parsons et al., 1991), b)  $FeO^{T}$ -Al<sub>2</sub>O<sub>3</sub> diagram (Abdel-Rahman, 1994) (C: biotites associated with subduction, P: biotites after continental collision, A: biotites formed in anorogenic regions), c)  $FeO^{T}$ -Al<sub>2</sub>O<sub>3</sub>-MgO triangle diagram (I: biotites associated with muscovite and topaz, II: biotites associated with other mafic minerals, III: biotites associated with hornblende, pyroxene, or olivine), d) MgO-FeO+MnO-TiO<sub>2</sub>\*10 triangle diagram (Speer, 1984) (A: primary brown biotites, B: primary re-equilibrated green and greenish-brown biotites, C: secondary green biotites).

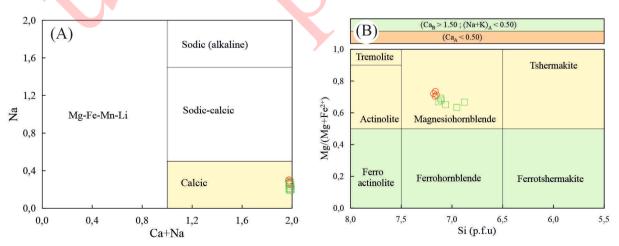


Figure 10- a-b) Composition of amphiboles in the studied plutonic rocks (Leake et al., 1997)

Rock type	Gabbr	oic diorit	e		Granodiorite							
Sample	S 13-1	S 13-2	S 13-3	IS 13-4	S 19-1	IS 19-2	S 19-3	IS 19-4	S 19-5	S 19-6		
(r-c)	(r)	(c)	(c)	(c)	(r)	(c)	(c)	(c)	(c)	(c)		
SiO <sub>2</sub>	49.83	50.24	49.61	50.21	47.64	47.14	48.87	48.14	48.65	49.12		
TiO <sub>2</sub>	0.77	0.89	1.07	0.83	1.30	1.15	0.80	0.88	1.13	0.83		
Al <sub>2</sub> O <sub>3</sub>	4.57	4.68	4.76	4.72	5.82	6.20	5.09	5.57	5.01	4.95		
FeO <sup>T</sup>	15.39	14.80	14.48	14.93	16.43	16.35	16.02	16.14	14.24	15.63		
MnO	0.53	0.54	0.47	0.51	0.54	0.55	0.49	0.47	0.43	0.57		
MgO	14.27	14.55	14.50	14.62	13.02	13.18	13.26	12.97	14.29	13.86		
CaO	11.09	11.15	11.17	10.99	11.15	10.90	11.27	11.34	11.40	10.97		
Na <sub>2</sub> O	1.22	1.12	1.13	1.14	1.50	1.50	0.97	1.05	1.30	1.27		
K <sub>2</sub> O	0.42	0.38	0.51	0.37	0.64	0.66	0.48	0.59	0.52	0.47		
Total	98.09	98.35	97.70	98.32	98.04	97.63	97.25	97.15	96.97	97.67		
Si	7.25	7.27	7.23	7.27	7.01	6.96	7.22	7.13	7.15	7.20		
Ti	0.08	0.10	0.12	0.09	0.14	0.13	0.09	0.10	0.12	0.09		
Al	0.78	0.80	0.82	0.80	1.01	1.08	0.89	0.97	0.87	0.85		
Fe <sup>(ii)</sup>	1.87	1.79	1.76	1.81	2.02	2.02	1.98	2.00	1.75	1.92		
Mn	0.07	0.07	0.06	0.06	0.07	0.07	0.06	0.06	0.05	0.07		
Mg	3.10	3.14	3.15	3.15	2.85	2.90	2.92	2.86	3.13	3.03		
Ca	1.73	1.73	1.74	1.70	1.76	1.72	1.78	1.80	1.79	1.72		
Na	0.69	0.63	0.64	0.64	0.86	0.86	0.56	0.60	0.74	0.72		
K	0.08	0.07	0.09	0.07	0.12	0.12	0.09	0.11	0.10	0.09		
Fe#	0.70	0.69	0.73	0.64	0.82	0.73	0.74	0.78	0.84	0.71		
Mg#	0.70	0.72	0.71	0.73	0.63	0.67	0.67	0.65	0.68	0.69		

Table 5- Microprobe analysis results of amphiboles.

4.3. Crystallization Conditions of Magma Forming the Kazıkbeli Pluton

The formation and stability of minerals within magma are primarily governed by variations in temperature (T), pressure (P), and oxygen fugacity (fO<sub>2</sub>). Determining these intensive variables (T, P, fO<sub>2</sub>) offers crucial insights into the crystallization history of magma. Thermodynamic principles enable the prediction of mineral phases that crystallize under specific conditions. Conversely, the mineral assemblage and composition within a rock can be utilized to infer the P-T conditions during its formation (thermobarometry). The accuracy of thermodynamic calculations hinges on the identification and propagation of errors throughout all stages of P-T estimation. Typically, thermobarometric calculations exhibit uncertainties of ±30-50°C for temperature and  $\pm 1-0.5$  kbar for pressure. Notably, the limited diversity of mineral assemblages encountered in plutonic

rocks presents a significant challenge for precise P-T determinations.

The ubiquity of amphibole-plagioclase pairs in calcalkaline plutonic rocks makes them a popular choice for thermobarometric calculations (Hammarstrom and Zen, 1986; Hollister et al., 1987; Johnson and Rutherford, 1989; Blundy and Holland, 1990; Schmidt, 1992; Holland and Blundy, 1994). Fortunately, the mineral assemblage of the studied plutons is amenable to this approach due to the presence of both amphibole and plagioclase. However, caution is warranted. Post-magmatic processes such as actinolite transformation, chloritization, and the formation of opaque minerals can compromise the integrity of hornblende compositions (Hammarstrom and Zen, 1986). Consequently, components exhibiting alteration must be meticulously excluded from thermobarometric calculations.

This investigation employed a suite of thermobarometric calculations to elucidate the pressure (P), temperature (T), oxidation state ( $fO_2$ ) conditions and water contents prevailing during the crystallization of the studied plutonic rocks. Mineral assemblages, including amphibole, plagioclase, biotite, K-feldspar, quartz, and Fe-Ti oxides, served as the basis for these calculations.

#### 4.3.1. Geobarometer

In order to constrain the crystallization pressures of the rocks from the Kazıkbeli Pluton, a multi-equilibrium approach was employed. This approach utilized established barometers, including the amphibole-Al<sup>(T)</sup> (Hammarstrom and Zen, 1986; Hollister et al., 1987; Johnson and Rutherford, 1989; Schmidt, 1992), amphibole (Ridolfi et al., 2010; Ridolfi and Renzulli, 2012), and biotite (Uchida et al., 2007) barometers were employed to constrain crystallization pressures. The results of these barometers are presented in Table 6 and detailed below.

Pressure estimation in plutonic rocks is often achieved through the analysis of Al<sup>(T)</sup> in amphiboles. This approach relies on the established positive correlation between Al<sup>(T)</sup> content and pressure, as documented by Hammarstrom and Zen (1986), Hollister et al. (1987), Johnson and Rutherford (1989), and Schmidt (1992). Various calibrations have been developed to convert Al<sup>(T)</sup> values into pressure estimates, each with specific applicability depending on the crystallization environment. For instance, the Johnson and Rutherford (1989) calibration (P (kbar) = 4.28 Al<sup>(T)</sup> - 3.54) is best suited for volcanic or highpressure plutonic settings. Conversely, the calibrations proposed by Hollister et al. (1987) (P (kbar) = 5.64Al<sup>(T)</sup> - 4.76) and Hammarstrom and Zen (1986) (P  $(kbar) = 5.03 \text{ Al}^{(T)} - 3.92)$  are more accurate for highpressure plutonic environments. Finally, the calibration by Schmidt (1992) (P (kbar) =  $4.76 \text{ Al}^{(T)} - 3.01$ ) is preferred for low-pressure settings. These variations in calibration effectiveness highlight the influence of the surrounding mineral assemblage (plagioclase, K-feldspar, biotite, pyroxene, magnetite, ilmenite, and potentially sphene) on the equilibrium relationships that govern Al<sup>(T)</sup> incorporation in amphiboles.

The estimation of crystallization pressures employed the aluminum-in-hornblende barometer technique applied to amphiboles, as detailed in Table 6. This approach yielded a range of pressures dependent on the specific calibration chosen. Hammarstrom and Zen (1986) proposed a pressure range expressed as 5.03  $AI^{(T)}$  - 3.92 kbar. Their calculations resulted in values between 0.07 and 0.15 kbar for gabbroic diorites and between 0.33 and 1.44 kbar for granodiorites (Table 6). Hollister et al. (1987) established a distinct pressure range of 5.64  $AI^{(T)}$  - 4.76 kbar, 0.04-0.08 kbar for gabbroic diorites and 0.11-1.25 kbar for granodiorites (Table 6). Finally, Schmidt (1992) presented a pressure estimation of 4.76  $AI^{(T)}$  - 3.01 kbar, translating to a range of 0.68 to 0.84 kbar for gabbroic diorites and 1.01 to 2.06 kbar for granodiorites (Table 6).

The pressure conditions under which the plutonic rocks crystallized were estimated using amphibole mineral chemistry data. Barometric formulas developed by Ridolfi et al. (2010) and Ridolfi and Renzulli (2012) (Table 6) were employed for these calculations. Ridolfi et al. (2010) based calculations resulted in a pressure range of 0.5–0.9 kbar, with distinct values of 0.5-0.6 kbar for gabbroic diorites and 0.6–0.9 kbar for granodiorites (Table 6). Similarly, the Ridolfi and Renzulli (2012) method yielded a pressure range of 0.6 to 1.0 kbar, with 0.6 to 0.7 kbar for gabbroic diorites and 0.7 to 1.0 kbar for granodiorites (Table 6).

Biotite crystallization pressures were determined using the empirical equation established by Uchida et al. (2007). This formula, calibrated based on Al<sup>(T)</sup> content obtained from mineral chemistry analyses, yielded a pressure range of 0.08-0.61 kbar for the studied plutonic rocks (Table 6). Notably, pressures varied from 0.08 to 0.18 kbar for gabbroic diorites and from 0.13 to 0.61 kbar for granodiorites (Table 6).

#### 4.3.2. Geothermometer

Amphibole-plagioclase (Holland and Blundy, 1994), amphibole (Ridolfi et al., 2010; Ridolfi and Renzulli, 2012), and biotite (Luhr et al., 1984) thermometers were utilized to estimate crystallization temperatures. The results obtained from these thermometers are presented in Table 7 and detailed below.

Magma crystallization temperatures were estimated using the Blundy and Holland (1990) amphibole-plagioclase thermometer. This approach relies on two equilibrium reactions involving edenite, tremolite, pargasite, albite, and quartz. The calculation employs a formula (T = 0.667P - 48.98 + Y / (-0.0429 - 0.008314 LnK)) that incorporates pressure (P) derived

from the assumption of equilibrium crystallization between amphibole and plagioclase. Within the formula, Y and K values are determined based on the composition of the amphibole (Xab, Si content). However, this specific thermometer is only applicable under certain conditions: (i) Equilibrium crystallization of amphibole-plagioclase must occur alongside quartz, K-feldspar, biotite, pyroxene, Fe-Ti oxide, and sphene. (ii) Plagioclase feldspar composition must be less calcic than An92 (albite molar fraction exceeding 0.08). (iii) Coexisting amphibole must have a cationic Si content below 7.8 atoms per formula unit. This type of amphibole-plagioclase thermometer is typically suited for estimating temperatures within the range of 500 to 1100 °C.

The calculated temperatures in the pluton, using the proposed formulas, range between 742-824°C for the studied rocks. Specifically, temperatures range from 742 to 744  $^{\circ}$ C for gabbroic diorites and from 757 to 824  $^{\circ}$ C for granodiorites (Table 7).

Equilibrium crystallization temperatures within the studied plutonic rocks were estimated using amphibole compositions and the thermometers developed by Ridolfi et al. (2010) and Ridolfi and Renzulli (2012) (Table 7). The Ridolfi et al. (2010) formulation yielded a temperature range of 738-797 °C for the entire pluton, with distinct values of 740-751 °C for gabbroic diorites and 738-797 °C for granodiorites (Table 7).

Amphibole compositions were employed to estimate equilibrium crystallization temperatures using the equations proposed by Ridolfi and Renzulli (2012). The calculated temperatures range from 712 to 804 °C throughout the pluton, with specific values of 744 to 748 °C for gabbroic diorites and 712 to 804 °C for granodiorites (Table 7).

Table 6- The pressure estimates calculated using amphibole-plagioclase, amphibole and biotite for the Kazıkbeli

	P (kbar)	P (kbar)	P (kbar)	P (kbar)	P (kbar)	P (kbar)
	(Hammarstrom		(Schmidt,	(Ridolfi et	(Ridolfi and	(Uchida et
	and Zen, 1986) Amphibole-Plagi	al., 1987)	1992)	al., 2010)	Renzulli, 2012)	al., 2007)
Minerals	Amphibole-Plagi	oclase	,	Amphibole		Biotite
Gabbroic diorite	1 0			1		
Min	0.07	0.04	0.68	0.50	0.60	0.08
Max	0.15	0.08	0.84	0.60	0.70	0.18
Avg	0.12	0.06	0.76	0.55	0.65	0.14
$SD(\pm)$	0.04	0.03	0.07	0.06	0.06	0.05
Min Depth (km)	0.3	0.1	2.5	1.9	2.2	0.3
Max Depth (km)	0.6	0.3	3.1	2.2	2.6	0.7
Granodiorite 🔺						
Min 🦰	0.33	0.11	1.01	0.60	0.70	0.13
Max	1.44	1.25	2.06	0.90	1.00	0.61
Avg	0.78	0.62	1.44	0.75	0.82	0.36
$SD(\pm)$	0.44	0.48	0.42	0.10	0.12	0.18
Min Depth (km)	1.2	0.4	3.7	2.2	2.6	0.5
Max Depth (km)	5.3	4.6	7.6	3.3	3.7	2.3
Min: Minimum, M	lax: Maximum, Av	/g: Average, SP (	(±): Standard	deviation.		

Depth was taken as 1kbar = 3.7 km for the continental crust (Tulloch and Challis. 2000).

Table 7- The temperature estimates calculated using amphibole-plagioclase, amphibole and biotite for the Kazıkbeli pluton.

	T (°C)	T (°C)	T (°C)	T (°C)
	(Blundy and Holland,	· · · · · · · · · · · · · · · · · · ·	`	(Luhr et al.,
	1990)	2010)	Renzulli, 2012)	1984)
Minerals	Amphibole-Plagioclase	Amphibole	Amphibóle	Biotite
Gabbroic diorite		*	*	
Min Max	742	740	744	768
Max	744	751	748	783
Avg	743	746	746	773
SD (±)	1	5	2	9
Avg SD (±) Granodiorite			_	
Min	757	738	712	721
Max	824	797	804	766
Avg	784	766	756	736
SD (±)	29	22	35	18

Biotite mineral chemistry analyses were employed to calculate crystallization temperatures within the studied plutonic rocks using the formulas proposed by Luhr et al. (1984) (Table 7). These calculations yielded a temperature range of 721-783 °C, with specific values of 768-783 °C for gabbroic diorites and 721-766 °C for granodiorites (Table 7).

#### 4.3.3. Oxygen Fugacity

Crystallization processes within magmatic systems are significantly influenced by oxygen fugacity (fO<sub>2</sub>), which represents the partial pressure of oxygen. This parameter exerts control over the pressuretemperature relationships governing melt behavior and dictates the stability fields of rock-forming minerals (Wones, 1989; Ridolfi et al., 2010). Notably, fO<sub>2</sub> exhibits a positive correlation with temperature, typically increasing as temperatures rise. Additionally, the oxygen content within silicate melts is influenced by the source of the heat driving the process and the rate of gas mixing (Wones, 1989). However, due to the slow cooling rates experienced by granitic magmas, direct determination of their original fO, values remains elusive. Consequently, researchers rely on relative approaches and calculations to estimate this crucial parameter (Wones, 1989; Anderson and Smith, 1995; Kemp, 2004).

Oxygen fugacity ( $\Delta$ NNO) within the studied plutonic rocks was estimated using amphibole compositions and the oxybarometry models developed by Ridolfi et al. (2010) and Ridolfi and Renzulli (2012) (Table 8). The Ridolfi et al. (2010) formulation yielded  $\Delta$ NNO values ranging from 1.3 to 2.0, with distinct values of 1.8-2.0 for gabbroic diorites and 1.3-1.7 for granodiorites (Table 8). In contrast, the Ridolfi and Renzulli (2012) model resulted in a  $\Delta$ NNO range of 0.2 to 1.0, with values of 0.4-0.6 for gabbroic diorites and 0.2-1.0 for granodiorites (Table 8).

Oxygen fugacity  $(fO_2)$  in the studied rocks was determined using the Wones (1989) method, which relies on pressure and temperature values. These pressure and temperature values were calculated using the approach outlined by Ridolfi et al. (2010). The calculations yielded a range of  $fO_2$  values from -13.8 to -12.5, with distinct values of -13.1 to -13.4 for gabbroic diorites and -12.5 to -13.8 for granodiorites (Table 8). Oxygen fugacity (fO2) values in the studied rocks were calculated using the temperature data derived from the method established by Wones (1989) (Table 8). These calculations yielded a range of fO2 values from -16.14 to -14.31, with distinct values observed in gabbroic diorites (-14.72 to -14.31) and granodiorites (-16.14 to -14.78) (Table 8).

#### 4.3.4. Water Content

The water content of magmas containing amphiboles remains a topic of debate, with reported values ranging from 2-3 wt% (Luhr et al., 1984) to a mean of 5 wt% (Eggler, 1972; Helz, 1973; Naney, 1983) or even 6 wt% (Merzbacher and Eggler, 1984). In this study, calculations based on the amphiboles within the studied samples yielded water content estimates between 4.4 and 7.8 wt%. The presence of hydrous mafic minerals (amphibole, biotite), titanite, and apatite in the studied plutons serves as an indicator of a magma rich in water and other volatile components. Notably, high-temperature magmas with such compositions may ascend to relatively shallow depths within the continental crust before complete crystallization (Helmy et al., 2004).

The water contents of amphiboles in the studied rocks were determined using the hydrometer formulas developed by Ridolfi et al. (2010) and Ridolfi and Renzulli (2012) (Table 8). Hydration states of amphiboles within the studied rocks were assessed using the hydrometer calculations proposed by Ridolfi et al. (2010) and Ridolfi and Renzulli (2012). The results are presented in Table 8.

Average water contents of amphiboles in the studied rocks, determined using Ridolfi et al. (2010) method, range from 3.7% to 4.5% (Table 8). Specifically, they vary between 3.9% and 4.3% for gabbroic diorites and 3.7% and 4.5% for granodiorites (Table 8).

Amphibole water contents within the studied rocks were determined using the method established by Ridolfi and Renzulli (2012) (Table 8). These calculations revealed an average water content range of 4.4% to 5.7%, 4.5% to 4.6% for gabbroic diorites. Granodiorites exhibited a wider range, varying between 4.4% and 5.7% (Table 8).

11011.						
	DNNO	DNNO DNNO		fO <sub>2</sub>	H <sub>2</sub> O	H <sub>2</sub> O
	(Ridolfi et	(Ridolfi and	(Ridolfi et	(Wones,	(Ridolfi et	(Ridolfi and
	al., 2010)	Renzulli, 2012)	al., 2010)	(1989)	al., 2010)	Renzulli, 2012)
Minerals	Amphibóle	, ,	, ,	Biotité	Amphibóle	, ,
Gabbroic d					1	
Min	1.80	0.40	-13.40	-14.72	3.90	4.50
Max	2.00	0.60	-13.10	-14.31	4.30	4.60
Avg	1.90	0.48	-13.20	-14.58	4.13	4.55
$SD(\pm)$	0.08	0.10	0.14	0.24	0.17	0.06
Granodiori	te		-			
Min	1.30	0.20	-13.80	-16.14	3.70	4.40
Max	1.70	1.00	-12.50	-14.78	4.50	5.70
Avg	1.52	0.52	-13.13	-15.67	4.17	4.83
$SD(\pm)$	0.16	0.46	0.47	0.54	0.31	0.51

Table 8- The oxygen fugacity and water content estimates calculated using amphibole and biotite for the Kazıkbeli pluton.

#### 4.4. Disequilibrium Parameters

The interplay between mineral compositions and textural features observed in rocks offers valuable insights into the various types of disequilibrium encountered during magma evolution. Factors such as cooling, decompression, and magma mixing can trigger variations in temperature, pressure, and composition, ultimately driving the magmatic system towards disequilibrium (Nixon, 1988; Rutherford and Hill, 1993; Simonetti et al., 1996; Perugini et al., 2003). Petrographic or textural criteria serve as key indicators of textural disequilibrium. These include the presence of sieved plagioclase crystals (Dungan and Rhodes, 1978), the coexistence of normal and sieved plagioclase within a single sample (Stimac and Pearce, 1992; Venezky and Rutherford, 1997), and the occurrence of rounded and embayed crystals (Stimac and Pearce, 1992). Compositional criteria provide another window into disequilibrium processes. The presence of both normally and reversely zoned crystals within the same sample serves as a diagnostic feature (Sakuyama, 1981). These variations in zoning patterns reflect the changing chemical composition of the magma as it undergoes disequilibrium.

Textural, petrographic, and mineral compositional features within the studied rocks provide compelling evidence for disequilibrium processes and compositional zoning, particularly in plagioclase (Figure 6). Oscillatory zoning observed in plagioclase crystals is a well-established indicator of disequilibrium (Figure 6d). Additionally, the occasional presence of sieve textures within these minerals offers further support for this interpretation. The occurrence of reverse zoning in some plagioclase grains, particularly noteworthy, and the coexistence of both normally and reversely zoned plagioclase

crystals within the same sample (Figure 11) constitute definitive evidence for disequilibrium crystallization. Furthermore, the presence of K-feldspar poikiloblasts enclosing plagioclase, amphibole, and opaque minerals (Figure 6) represents another textural feature suggestive of disequilibrium crystallization.

Normal zoning patterns in plagioclase crystals are readily attributable to fractional crystallization processes. However, the origin of reverse zoning remains a topic of debate, with several competing explanations: (i) Blundy and Cashman (2001) propose that reverse zoning can arise due to an increase in both temperature and pressure within a water-saturated magma system. (ii) Alternatively, Blundy et al. (2006) suggest that reverse zoning may develop during the ascent of water-saturated magmas as a consequence of rising temperatures. (iii) Couch et al. (2001) and Streck (2008) posit that back-mixing of cooler magmas with hotter, inflowing magmas can induce temperature increases, leading to reverse zoning in plagioclase. The studied samples exhibit a wide compositional range (An26-66) within the rims of individual plagioclase crystals (Table 2). This significant compositional variability within a single sample is consistent with a complex origin for the plagioclase, likely influenced by magma mixing processes (Wallace and Carmicheal, 1994).

#### 4.5. Evolution of Magma within the Crust

The Eastern Pontide (EP) has predominantly experienced a compressional tectonic regime throughout its geological history, particularly intensifying since the Paleozoic-Mesozoic transition. This compression resulted in the development of fracture systems oriented in the NE-SW and NW-SE directions. Notably, the elongated axes of plutons

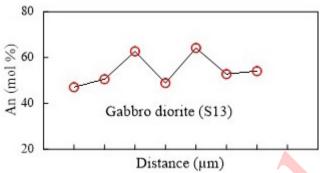


Figure 11- The observed variations in An (%) of zoned plagioclase minerals in the studied pluton (Symbols are as in Figure 4).

within the EP exhibit a strong correlation with these primary fracture trends. Gedikoğlu (1979) first documented this observation, and subsequent studies in the region have further corroborated this relationship (Kaygusuz et al., 2008, 2012, 2016, 2020, 2021).

Field observations reveal that the Kazıkbeli Pluton exhibits an elongated, elliptical shape and is situated subparallel to regional NE-SW trending faults. This geometry suggests emplacement within a regional zone of tectonic weakness. The pluton exhibits sharp, incompatible contacts with the surrounding finegrained country rocks. Notably, these country rocks occasionally display porphyritic and granophyric textures, and in some locations, xenoliths derived from the country rocks have been incorporated into the pluton. These combined observations, including the shape, contact relationships, and presence of xenoliths, are indicative of emplacement at relatively shallow crustal depths.

The presence of Early Eocene aged (~ 55 Ma) adakitic rocks in the Eastern Pontides corresponds to the last stage of the arc-continent collision and is associated with a collisional or post-collisional origin (Topuz et al., 2005, 2011; Karslı et al., 2010, 2011). In the Middle Eocene ( $\sim$  46-40 Ma), the presence of I-type, metaluminous, medium to high-K and occasionally shoshonitic plutonic rocks are linked to collisional eventsand subsequent geodynamic processes. Preexisting studies (Boztuğ et al., 2004; Topuz et al., 2005; Arslan and Aslan, 2006; Karslı et al., 2007; Kaygusuz et al., 2011; Arslan et al., 2013; Aydınçakır, 2014; Kaygusuz and Öztürk, 2015; Yücel et al., 2017; Temizel et al., 2018) indicated that the geochemical signatures are consistent with derivation from enriched lithospheric mantle or lower continental crust, or a mixture of both. The Eocene magmatism is attributed to a tensional regime associated with collision, post-collisional crustal thickening, and lithospheric delamination (Kaygusuz et al., 2011; Karslı et al., 2012; Temizel et al., 2012; Arslan et al., 2013; Aslan et al., 2014; Yücel et al., 2017). Lithospheric delamination following slab break-off is proposed as a mechanism for the formation of the Kazıkbeli plutonic rocks. The developing delamination enabled the upward movement (local or regional) of the asthenosphere and provided a hot asthenosphere thermal anomaly, leading to the partial melting of the chemically enriched lithospheric mantle and the formation of magmas that led to the formation of Middle Eocene rocks. After that, thinning and fracture systems formed in the crust due to tectonic extension allowed these melts to move upwards within the crust. During magma ascent, a chamber formed at a depth of approximately 3-5 km where fractional crystallization and magma mixing events occurred. Barometer estimates based on mineral assemblages and petrographic features (Table 6) suggest these magma chambers existed at depths ranging from 1 to 8 km (Figure 12). Magma mixing is further supported by textural features such as sieve texture and zoning. Finally, estimates of oxygen fugacity indicate oxidizing conditions prevailed during crystallization at this depth.

#### 5. Discussion

The amphibole-plagioclase and amphibole barometers employed in this study yielded comparable pressure values ranging from 0.04 to 2.06 kbar (Table 3). While the pressures estimated from the biotite barometer were slightly lower (0.08-0.61 kbar), they remained within the same general range. These findings are consistent with observations reported in the literature, where pressure values obtained using the hornblende-Al barometer for plutons align well with geological features and emplacement depths (Tulloch and Challis, 2000). In light of the convergent pressure estimates obtained from various methods and the qualitative observations from the studied plutons, the emplacement pressure can be confidently constrained to a range of approximately 1-2 kbar. Notably, the calcic nature of the amphiboles in the pluton, characterized by Al<sup>(T)</sup> values below 2.0, further supports a shallow intrusion environment (Hammarstrom and Zen, 1986). Textural features observed within the plutons, such as the presence of porphyritic textures and mineral regrowth (Figure 6), also lend credence to a shallow emplacement depth.

Solidus temperatures, typically exceeding

700°C, are often estimated using thermometers like the amphibole-plagioclase method (Anderson, 1996). The amphibole-plagioclase thermometer (Holland and Blundy, 1994) in this study gave temperature estimates within a narrow range (742-824 °C) and similar to the amphibole thermometer (738-804 °C) (Table 7). Additionally, the biotite thermometer (Luhr et al., 1984) yielded lower temperature estimates (721-783 °C) (Table 7). However, these low biotite-derived temperatures likely reflect re-equilibration with subsolidus intracrystalline variations during feldspar cooling (Toksoy-Köksal, 2016). The broad temperature range obtained from the hornblende-plagioclase thermometer may also suggest re-equilibration processes in slowly cooling plutonic rocks.

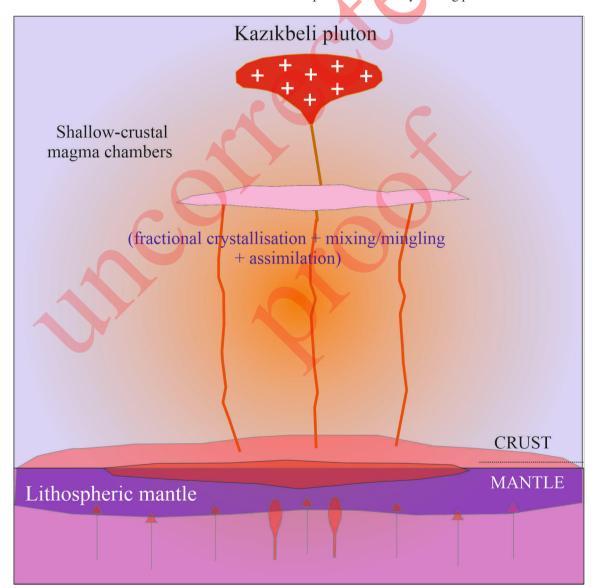


Figure 12- Schematic cross-section of the studied Kazıkbeli pluton within the crust, showing its emplacement and development.

Mineral chemistry plays a critical role in establishing the emplacement depth of plutonic rocks. Microprobe analyses of amphiboles from the Kazıkbeli Pluton (Table 5) reveal compositions that plot within the low-pressure field of Figure 13. This geochemical signature is indicative of crystallization at relatively shallow depths within the Earth's crust.

Pressure estimates derived from amphiboleplagioclase, amphibole, and biotite barometry in the studied plutonic rocks range from 0.04 to 2.06 kbar (Table 6). These values are comparable to pressures reported for other Eocene plutons in the Eastern Pontide: 1-3.8 kbar for the Sariçiçek and Dölek plutons (Karslı et al., 2007) and 0.1-2.8 kbar for the Kemerlikdağı, Aydıntepe, and Pelitli Plutons (Kaygusuz et al., 2020). However, pressures in the studied plutons are generally lower compared to a broader range documented for the Sariçiçek, Dölek, Üzengili, Arslandede, and Sorkunlu plutons (0.3-8.2 kbar; Eyuboğlu et al., 2017). Supporting the interpretation of shallow emplacement depths are the compositional characteristics of the amphiboles. These amphiboles are classified as calcic, with Al<sup>(T)</sup> values typically less than 2.0. According to Hammarstrom and Zen (1986), such low Al<sup>(T)</sup> contents are indicative of crystallization at shallow crustal levels. Furthermore, textural features observed in the studied rocks, such

as the presence of graphic intergrowths, further corroborate the shallow emplacement depths.

Thermometry calculations utilizing amphibole, amphibole-plagioclase, and biotite compositions in the investigated plutonic rocks yielded temperature values ranging from 712 to 824 °C (Table 7). This temperature range is consistent with values reported for other plutons of similar age within the Eastern Pontide: 617-768 °C for the Sariçiçek and Dölek plutons (Karslı et al., 2007) and 616-1172 °C for the Kemerlikdağı, Aydıntepe, and Pelitli Plutons (Kaygusuz et al., 2020). However, the temperature range for the studied plutons falls within a narrower window compared to the broader range documented for the Sariçiçek, Dölek, Üzengili, Arslandede, and Sorkunlu plutons (388-1196 °C; Eyuboğlu et al., 2017).

It is well established that thermometers like the amphibole-plagioclase method typically yield temperatures exceeding the solidus temperature (typically >700°C) (Anderson and Smith, 1995). In the case of the studied plutonic rocks, the temperatures derived from the amphibole-plagioclase thermometer fall below the expected solidus value. This observation suggests that the amphiboles may have undergone sub-solidus re-equilibration processes, as described by Moazzen and Droop (2004).

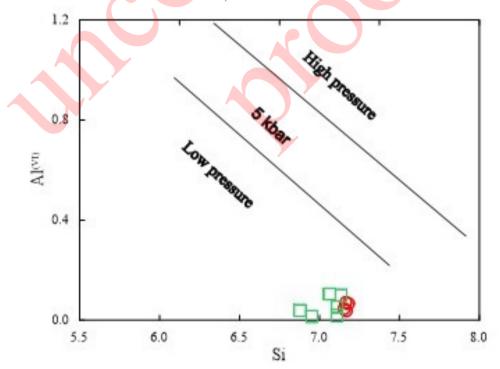


Figure 13- Composition and pressure range of amphiboles in the studied Kazıkbeli Pluton.

Biotite compositions within the studied plutonic rocks exhibit characteristics suggestive of a potential mantle origin. These biotites display intermediate Al<sup>4</sup> content (2.18-2.36 formula units per biotite; f.u.b.) and partially high Mg# values (0.52-0.58) (Figure 14). These compositional features partially resemble biotites documented in plutons with established mantle sources, which typically exhibit Al<sup>4</sup> contents of 2.3-2.4 f.u.b. and Mg# exceeding 0.60.

Direct determination of the original oxygen fugacity ( $fO_2$ ) in granitic magmas is challenging due to the effects of post-crystallization processes during gradual cooling (Wones, 1989; Anderson and Smith, 1995; Kemp, 2004). As a consequence, researchers rely on relative approaches and calculations to estimate  $fO_2$  values. In the studied plutonic rocks, the calculated log<sub>10</sub>  $fO_2$  values range from -12.5 to -16.14. Notably, these values exhibit good agreement with those reported for the Eocene Sariçiçek and Dölek plutons (-15 to -21; Karsh et al., 2007).

The presence of hydrous mafic minerals (amphibole) alongside apatite and titanite in the

studied rocks suggests a magma enriched in water and other volatile components. Such high-temperature, water-rich magmas are capable of ascending to shallow crustal depths while remaining partially molten (Helmy et al., 2004). The water content of amphibole-bearing magmas remains a topic of debate. While Luhr et al. (1984) reported values around 2-3%, other studies suggest an average closer to 5% (Eggler, 1972; Helz, 1973; Naney, 1983), with Merzbacher and Eggler (1984) proposing an average of 6%. Notably, the water content estimates derived from the analyzed amphiboles in the studied plutons fall within a range of 3.7 to 5.7 wt%.

Thermobarometric calculations utilizing mineral compositions revealed a wide range of pressure (0.04-2.06 kbar) and temperature (712-824 °C) estimates within the studied plutonic rocks (Tables 6-7). These combined observations suggest the emplacement of a water-rich magma derived from the continental crust at relatively shallow depths within the crust, likely in the range of  $\sim$ 1 to 8 km.

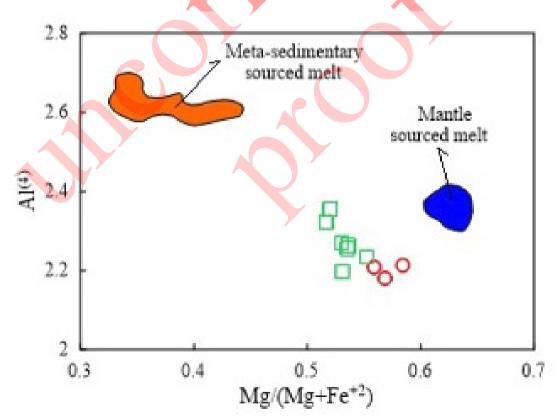


Figure 14- The Al<sup>4</sup> versus Mg/(Mg+Fe+2) diagram for biotites in the studied rocks used to distinguish between two end-member melt fields: meta-sedimentary (Kemp, 2001) and mantle-derived (Kemp, 2004).

#### 6. Results

The Kazıkbeli Pluton is an elongated, elliptical intrusion oriented in a NE-SW direction, encompassing an area of approximately 46 km<sup>2</sup>. It exhibits lithological variability, ranging from gabbroic diorite to granite. The rocks of the pluton are composed of plagioclase (An<sub>26-66</sub>), orthoclase (Or<sub>80-97</sub>), quartz, amphibole, biotite, and Fe-Ti oxides.

The Kazıkbeli rocks display a textural spectrum, encompassing fine- to medium-grained, porphyritic, poikilitic, and occasionally microlitic and graphic textures. Amphibole compositions vary in Mg# (0.63-0.73), and biotites exhibit Mg# values ranging from 0.52 to 0.58. Calculated crystallization temperatures derived from amphibole and biotite data range from 712°C to 824°C. Pressure estimates suggest emplacement at relatively shallow crustal depths (~1 to 8 km), with values ranging from 0.04 to 2.06 kbar. Oxygen fugacity ( $fO_2$ ) values exhibit minimal variation, falling within a range of -12.5 to -13.8. Water contents estimated from amphiboles range between 3.7% and 5.7%.

In light of the combined geochemical and textural data, the Kazıkbeli Pluton is interpreted to have solidified at relatively shallow crustal depths, likely within a range of  $\sim$ 1 to 8 km.

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